Methodological interference of biochar in the determination of extracellular enzyme activities in composting samples

K. Jindo1,3, K. Matsumoto2, C. García Izquierdo1, T. Sonoki2, and M. A. Sanchez-Monedero1

1Centro de Edafología y Biología Aplicada del Segura (CEBAS-CSIC), Department of Soil Conservation and Waste Management. Campus Universitario de Espinardo, 30100 Murcia, Spain
2Faculty of Agriculture and Life-Sciences, Hirosaki University, Bunkyo-cho, Hirosaki, Aomori 036-8561, Japan
3Institute of Industrial Science, the University of Tokyo, 4-6-1 Komaba Meguro-Ku, Tokyo 153-8505, Japan

Correspondence to: K. Jindo (keijindo@hotmail.com)

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Abstract. Biochar application has received increasing attention as a means to trap recalcitrant carbon and enhance soil fertility. Hydrolytic enzymatic assays, such as β-glucosidase and phosphatase activities, are used for the assessment of soil quality and composting process, which are based on use of p-nitrophenol (PNP) derivatives as substrate. However, sorption capacity of biochar can interfere with colorimetric determination of the hydrolysed PNP, either by the sorption of the substrate or the reaction product of hydrolysis into biochar surface. The aim of the present work is to study the biochar sorption capacity for PNP in biochar-blended composting mixtures in order to assess its impact on the estimation of the colorimetric-based enzymatic assays. A retention test was conducted by adding a solution of known amounts of PNP in universal buffer solution (pH = 5, 6.5 and 11, corresponding to the β-glucosidase, acid and alkaline phosphatase activity assays, respectively), in samples taken at the initial stage and after maturation stage from four different composting piles (two manure composting piles; PM: poultry manure, CM: cow manure and two other similar piles containing 10% of additional biochar (PM + B, CM + B)). The results show that biochar-blended composts (PM + B, CM + B) generally exhibited low enzymatic activities, compared to manure compost without biochar (PM, CM). In terms of the difference between the initial and maturation stage of composting process, the PNP retention in biochar was shown higher at maturation stage, caused most probably by an enlarged proportion of biochar inside compost mixture after the selective degradation of easily decomposable organic matter. The retention of PNP on biochar was influenced by pH dependency of sorption capacity of biochar and/or PNP solubility, since PNP was more efficiently retained by biochar at low pH values (5 and 6.5) than at high pH values (11).

1 Introduction

Agricultural use of biochar has been receiving attention as an alternative strategy for mitigation of greenhouse gas (GHG) emission as well as improvement of soil properties. In addition, high sorption character of biochar, similarly to activated carbon, makes it possible to contribute to reduction of several hazards (heavy metals, pesticide, and hydrocarbon) in soil (Yang et al., 2009). Furthermore, the suitability of biochar has been reported as an additional component for enhancing the composting quality by reducing the nitrogen volatilization due to sorption on surface of biochar (Steiner et al., 2010), mitigating CH₄ emission due to the higher aeration in composting pile (Sonoki et al., 2012) and improving compost quality such as an intense humification process and more recalcitrant character (Dias et al. 2010; Jindo et al., 2012). Lately, the application of biochar-blended compost to soil can promote a synergistic effect on enhancing plant nutrition content and water-holding capacity (Lieu et al.2012) as well as contributing the immobilization of organic pollutants and heavy metal (Beesley et al., 2010).

In terms of the decomposition of organic matter during composting, enzymatic activities such as β-glucosidase and phosphatase are a useful tool to reflect dynamics of biodegradation process and provide valuable information about the
stability and maturity of the compost (Vuurinen 2000; Mondini et al., 2004). The measurement of enzymatic activities is utilized not only for the composting process but also as an indicator of soil quality since they are involved in the dynamics of soil nutrient cycle (Jordan et al., 1995). These hydrolytic enzymes are measured by colorimetric determination of \( p \)-nitrophenol (PNP) which is formed as the reaction product of hydrolysis of different nitrophenyl derivatives used as a substrate: nitrophenyl-\( \beta \)-d-glucopyranoside (PNG) for \( \beta \)-glucosidase activity, and \( p \)-nitrophenyl phosphatase (PNPP) for alkaline and acid phosphatase activities. By contrast, \( p \)-nitrophenol is a well-known toxic compound in industrial sector, and is treated by absorption on activated carbon (Tang et al., 2007; Ivančev-Tumbas al 2008). Furthermore, some biochar, produced at high temperature, has similar absorption character to activated carbon (Hale et al., 2013) and interferes with the extraction of soluble organic compounds, leading to underestimation of soil microbial activities (Chan et al., 2007). Even though several works on the relation between microbial measurements and biochar exposure have been reported (Durenkamp et al., 2010; Bailey et al., 2011; Luo et al., 2013), further research is required for understanding the biochar interaction from the chemical, physical and biochemical point of view. Thies and Rillig (2009) proposed the utilization of spiking assays with specific molecules as internal standard to overcome potential interferences in the estimation of the microbial parameters.

The aim in present work was to study the influence of biochar as a composting component on the retention of the PNP generated from three colorimetric-based enzymatic assays (alkaline and acid phosphatases and \( \beta \)-glucosidase). The retention of PNP was tested in two different composting mixtures (poultry manure (PM) and cow manure (CM)) and other two similar composting mixtures containing biochar as an additional component (PM + B, CM + B).

2 Materials and methods

2.1 Biochar preparation

The production of biochar, made from broad-leaved tree (\textit{Quercus serrata} Murray), was carried out using a Japanese traditional kiln at atmospheric pressure and a temperature range of 400–600 °C with a final temperature of 550 °C. To analyse the physical properties of biochar, we ground and sieved the biochar to less than 0.5 mm in diameter. The elemental content was measured with an elemental analyzer (Thermo Finnigan EA1112, Thermo Fisher Scientific, Inc., MA, USA). The pH was measured with a compact pH meter B-212 (HORIBA Ltd., Kyoto, Japan). Microporosity was evaluated by the iodine (I2) number method, and methylene blue (MB) adsorption capacity was measured, following the initial methodology used by Gaspard et al. (2007). Surface area was measured with a BELSORP18PLUS (BEL Japan, Inc., Osaka, Japan). The main characteristics of the obtained biochar are shown in Table 1.

2.2 Raw materials and composting process

Composting was carried out at Kanagi experimental farm of Hirosaki University. Two composting mixtures were prepared following initial proportion of organic waste: CM – cattle manure (100.9 kg) mixed with apple pomace (76.8 kg), rice straw (9.7 kg) and rice bran (12.7 kg); PM – poultry manure (35.2 kg) mixed with apple pomace (141.8 kg), rice straw (9.9 kg) and rice bran (13.0 kg). Another two composting mixtures (CM + B and PM + B) were prepared by enriching the initial composting mixtures CM and PM with 20 kg of biochar. The organic waste mixtures were composted in cone shaped windrows with regular turnings and continuous monitoring of pile temperature and moisture. The principal physicochemical properties of the composting mixtures are described in Table 2, and further information on the composting process and characteristics of the composting mixtures can be found elsewhere (Sonoki et al., 2012). The composting process lasted approximately 3 months for all piles. A representative sample of each organic material was taken at the initial stage (I) and after maturation stage (M). These samples were collected from different spots of piles, mixed together, air dried and ground to 0.5 mm.
2.3 Thermogravimetric analysis (TGA)

Thermal analysis of the organic material was measured using a SDT-2960 simultaneous DSC-TGA thermal analyzer (TA instruments) under static air atmosphere as follows: a temperature equilibrating at 30°C followed by a linear heating rate of 5°C min⁻¹ from 30 to 105°C, an isotherm for 10 min and then continued ramping of 5°C min⁻¹ from 105 to 680°C. An index of thermal lability of the organic matter (W₂/W₁), shown in Table 2, was calculated from the ratio of mass loss at 350–550°C (W₂)/mass loss at 110–350°C (W₁) (Plante et al., 2009).

2.4 Enzymatic analysis

Alkaline and acid phosphatase and β-glucosidase activities were determined following the methods reported by Tabatabai and Bremmer (1971) and Eivazi and Tabatabai (1988), respectively, using 0.5 g of organic material, and 2 mL of modified universal buffer (MUB) containing the following substrate: alkaline phosphatase activity assay was performed at pH 11 using p-nitrophenyl phosphatase (PNPP) as substrate. Meanwhile, acid phosphatase activity assay was performed with the same substrate at pH 5.5; β-glucosidase activity was assayed at pH 6 using p-nitrophenyl β-D-glucopiranoside (PNG) as substrate. In the three cases, the suspensions were incubated at 37 °C for 1 h. Enzymatic reactions were stopped by cooling in ice for 15 min. Then, 0.5 mL of CaCl₂ 0.5 M and 2 mL of NaOH 0.5 M (for phosphatases) or 2 mL of Tris (hydroxymethyl) aminomethane–sodium hydroxide (THAM-NaOH) 0.1 M pH 12 (for β-glucosidase) were added. After the centrifugation and filtration, the p-nitrophenol (PNP), formed as product reaction from the three enzymatic assays, was determined at 398 nm using a spectrophotometer.

2.5 PNP retention assay during the enzymatic activity analysis

To study the retention of PNP during the analysis of the different enzymatic activities, the following spiking assay was performed: instead of adding the substrates (PNG and PNPP) at the beginning of the procedure, reaction product (PNP) was added with different concentration (0, 50, 100 and 150 mg L⁻¹) to buffer solution (pH = 5, 6, 5 and 11, corresponding to the β-glucosidase, acid and alkaline phosphatase activity assays, respectively). After the incubation, the same procedure as for enzymatic assay, described in the previous section took place for PNP determination. This procedure allows evaluating the retention of PNP by the biochar during the analysis. Controls were performed similarly by adding the same amounts of PNP after the incubation period and before the measurement of the absorbance in the calibrated spectrophotometer (with an external PNP standard solution). These results are shown in Fig. 2. (CM and CM + B) and Fig. 3. (PM and PM + B). Lately, the PNP retention was calculated by fitting the amount of PNP measured after the enzymatic determination (PNPexp) and the amount

Table 1. Chemical and physical property of hardwood biochar (from broad-leaved tree (Quercus serrata Murray)).

<table>
<thead>
<tr>
<th>C</th>
<th>O</th>
<th>H</th>
<th>N</th>
<th>P</th>
<th>K</th>
<th>pH</th>
<th>MB absorption capacity</th>
<th>Iodine number</th>
<th>Surface area</th>
</tr>
</thead>
<tbody>
<tr>
<td>g kg⁻¹</td>
<td>-</td>
<td>mg kg⁻¹</td>
<td>mg kg⁻¹</td>
<td>m² g⁻¹</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>791.5</td>
<td>91.5</td>
<td>18.9</td>
<td>37.6</td>
<td>2.3</td>
<td>14.1</td>
<td>7.2</td>
<td>8.3</td>
<td>100</td>
<td>255.0</td>
</tr>
</tbody>
</table>

Figure 2. Retention of PNP with different pH solution in the compost samples of poultry manure PM (square), poultry manure + biochar PM + B (circle), poultry manure control PM-C (triangle), and poultry manure + biochar control PMB-C (diamond) during the composting process.

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Table 2. Physical and chemical properties in poultry manure compost (PM), poultry manure blended with biochar (PM + B), cow manure (CM), and compost blended with biochar (CM + B).

<table>
<thead>
<tr>
<th>Origin</th>
<th>Initial stage</th>
<th>Maturity stage</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>C (%)</td>
<td>N (%)</td>
</tr>
<tr>
<td>PM</td>
<td>36.9</td>
<td>1.7</td>
</tr>
<tr>
<td>PM + B</td>
<td>39.9</td>
<td>1.7</td>
</tr>
<tr>
<td>CM</td>
<td>35.6</td>
<td>1.7</td>
</tr>
<tr>
<td>CM + B</td>
<td>37.1</td>
<td>1.3</td>
</tr>
</tbody>
</table>

1 O.M. = organic matter.  
2 $W_2/W_1$ is the main weight losses occurred in the 110 to 350 °C ($W_1$) and 350 to 550 °C ($W_2$) ranges.

3 Results and discussion

3.1 Characteristics of the composting mixtures

Different composting mixtures were selected at different stages of the composting process to cover the range of organic matter stabilization degree. The different nature of the organic matter at different stages of the biodegradable process and the property of the recalcitrant biochar was assessed by thermogravimetry (Lyons et al., 2006; Tsui and Juan, 2010; Manya et al., 2013). Basically, the TGS-DSC diagrams are characterized by two main mass losses, showing two exothermic peaks, and these correspond respectively to the volatilization of light compounds such as aliphatic molecules or carbohydrates and another to the oxidation of high molecular weight components (Fig. 1). Comparing the graph shapes between the samples from initial stage (Fig. 1a and c) and from maturity stage (Fig. 1b and d), the second wave of peak, generated by mass loss at 350–550 °C, was pronouncedly shown at maturity stage, due to the selective degradation of labile organic materials during the composting process. As a consequence, the index of lability of $W_2/W_1$ in all samples at the maturity stage is higher than those at the initial stage (Table 2).

The influence of adding biochar into the composting mixture at the initial phase (Fig. 1a and c) is observed by the higher peak of second wave in biochar-blended composts (PM + B, CM + B), which are described in dotted lines (Fig. 1a and c). This is because biochar that has resulted from biochar originated from hardwood mostly consists of recalcitrant compounds, which are combusted at $W_2$ range (350–550 °C) in an oxidant atmosphere of air. Consequently, $W_2/W_1$ ratio at initial time (Table 2) increased in biochar-blended piles (PM + B, CM + B) from the piles of non-biochar addition (PM, CM). After maturation stage (Table 2), $W_2/W_1$ ratio markedly increased in the biochar-blended composts (PM + B, CM + B, 2.3, and 1.6, respectively), reflecting the high relative proportion of recalcitrant biochar.
Table 3. Percentage of PNP recovery calculated by the linear fitting of the measured amount of PNP (PNP_{exp}) and the amount of PNP added (PNP_{add}), according to the equation PNP_{exp} = k \times PNP_{add}. The percentage of PNP recovery is expressed as 100 \times k. Poultry manure compost (PM), poultry manure blended with biochar (PM + B), cow manure (CM), and cow manure blended with biochar (CM + B).

<table>
<thead>
<tr>
<th>Enzymatic assay</th>
<th>Poultry manure</th>
<th>Cow manure</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Initial stage</td>
<td>Maturity stage</td>
</tr>
<tr>
<td></td>
<td>PM</td>
<td>PM + B</td>
</tr>
<tr>
<td>Alkaline Phosphatase (pH 11)</td>
<td>88.9</td>
<td>78.6</td>
</tr>
<tr>
<td>Acid Phosphatase (pH 6.5)</td>
<td>84.6</td>
<td>71.3</td>
</tr>
<tr>
<td>β-glucosidase (pH 5)</td>
<td>72.7</td>
<td>59.0</td>
</tr>
</tbody>
</table>

* Standard error in parentheses.

3.2 Study of the PNP retention on biochar-blended compost

The colorimetric determination of PNP was influenced by the degree of stability of the composting mixtures, which affected the relative proportion of biochar in the mixture. The biochar-blended composts showed more retention of PNP, especially in the case of maturity stage (Figs. 2 and 3). The amount of PNP retained by the biochar-blended composting mixtures (CM + B and PM + B) varied from 41 % in the starting composting mixtures up to 74 % in mature composts. This result might have been attributed to gained dominance of biochar amount inside composting mixtures which was gradually increased during the composting process. The recalcitrance of biochar character was retained until the maturation stage, while labile organic materials in the composting piles were lost due to the selective degradation, as already shown by TGS measurement (Fig. 1). Therefore the effect of the physico-chemical properties of biochar on the compost structure is expected to be more dominant in the mature stage than at the initial stage.

The PNP retention by biochar also depends on pH status of the buffer solution, used by each specific enzymatic activity. At high pH condition (pH 11), representing alkaline phosphatase essay, the PNP retention is observed in the range between 15 and 30 % of the added PNP (Table 3). However, the same spiking assays performed at low pH (pH 6.5 and 5 from acid phosphatase and β-glucosidase activities, respectively) exhibited high PNP retention from 30 % (acid phosphatase determination in PM + B) up to 70 % which is the case of the β-glucosidase determination in CM + B. These results are in agreement with the pH dependence of phenol adsorption efficiency by activated carbon reported by several authors (Ayranci and Duman 2005; Tang et al., 2007), concluding that the absorption efficiency of activated carbon is lower in alkaline solution than neutral or acid solution. An increase in the amount of OH ions in alkaline solution reduces the diffusion of phenol ions due to an electrostatic repulsion of negatively charged site of the sorbent and phenolic ions. As the pH increases, the surface charge of pyrogenic materials became negative and decreases its sorption capacity (Beker et al., 2010). Furthermore, other authors (Zhang et al., 2010) reported that, regarding the mobility of biochar particles, the lower the pH solution, the lesser transport of the biochar particles.

The sorption affinity of pyrogenic material is also influenced by physical properties such as microporosity and surface area, as well as chemical properties such as hydrophobicity in relation with O/C content (Al-Asheh et al., 2004; Ko et al., 2007; Tsui and Juang, 2010). Micropore and mesopore structure, estimated respectively by the iodine number and the methylene blue adsorption, are usually enlarged at high pyrolysis temperature together with surface area. Overall, all these biochar properties are dominantly defined by feedstock and pyrolysis conditions used for the preparation of the biochar (Uchimiya et al., 2010).

The PNP retention by the organic matter of the composting mixtures prepared without biochar (PM and CM) was also affected by the pH gradient. Table 3 shows that, at low pH solution (pH 5, initial stage of composting), CM has 69 % of PNP recovery, meaning 31 % of PNP was retained. This methodological problem in the determination of the enzymatic activities is well-known in clay mineral soils or soils enriched with organic matter (Tabatabai and Bremer 1971; Trasar-Cepeda et al 1988). Organic material containing a large amount of humic substances is known to easily absorb PNP molecules (Chen et al., 2009). In order to tackle this obstacle, several authors have recently recommended testing the soil enzymatic assays in samples blended with biochar to ensure the assumption of saturating substrate concentrations, and if necessary amending the protocols before initiating the assays (Swine et al., 2013). In practice, and in order to overcome the underestimation by absorption on biochar, Paz-Ferreiro et al. (2012, 2014) used different calibration curves for each different type of amendment to acquire an accurate
measure of soil enzymatic activities. This problem is even more complex in composting samples, where the degradation of labile organic matter causes a progressive enrichment in the proportion of biochar in the mixture. The different proportion of biochar in the starting mixtures and the mature compost also requires the adaptation and optimization of the enzymatic assay to the different composting stage.

In conclusion, the presence of biochar limited the validity of enzymatic essays for the colorimetric determination of PNP since PNP was strongly retained in biochar-blended compost. It is a challenge to improve the colorimetric methods of PNP determination for biochar interaction, and a clear-cut solution has not been found yet. Further research is necessary in order to measure with colorimetric methods potential enzymatic activity in the presence of biochar.

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